Impact of Different Pollutant Sources on Human Health Using Solid Aerosol's Elemental Analysis

Muhammad Attique Khan Shahid^{1,*}, Khadim Hussain² and Ahmad Raza¹

¹Department of Physics, G.C. University, Faisalabad, Punjab, Pakistan

²Department of Physics, High Energy Physics, Punjab University, Lahore, Punjab, Pakistan

Abstract: Atomic absorption spectroscopy (AAS) was used in this study to find out the metal constituents and concentration for Seven (7) trace metals in the atmosphere of Faisalabad. The maximum elemental constituents and concentration for Zn, Cu, Cr, Ni, Pb, Mg and Cd in ppm were found to be Cadmium Cd(1) in amounts ranging between ($(2.596 \rightarrow 1.95475)$) in Pool ($(5 \rightarrow 3)$, Chromium Cr(2) in amounts ranging between ($(0.0145 \rightarrow 0.01125)$) in Pool ($(2 \rightarrow 3)$, Nickel Ni(3) in amounts ranging between ($(0.9925 \rightarrow 0.35575)$) in Pool ($(5 \rightarrow 3)$, Lead Pb(4) in amounts ranging between ($(1.33675 \rightarrow 0.2632)$) in Pool ($(2 \rightarrow 3)$, Zinc Zn(5) in amounts ranging between ($(2.515 \rightarrow 1.38825)$) in Pool ($(4 \rightarrow 5)$), Magnesium Mg(6) in amounts ranging between ($(1.22125 \rightarrow 1.15875)$) in Pool ($(4 \rightarrow 5)$), Calcium Ca(7) in amounts ranging between ($(1.46725 \rightarrow 3.53875)$) in Tool ($(4 \rightarrow 3)$) respectively. Following pool wise trend pattern of identified elements in solid aerosols is given in Table 1 & 2. The comparison of results reported in literature with the obtained results showed some differences in concentrations which could be explained on the basis of climatological and meteorological set up of difference pools under investigations. Furthermore, the health hazards due to identified trace metals were also investigated and were found that the metals were highly toxic and generating serious health hazards.

Keywords: Elemental atmospheric air pollution, co-relationship with health hazards, confirmed *via* Empirical relations and sociological survey need of protective measures.

1. INTRODUCTION

wider includes Environment, in its sense. everything, which is external to a human being. Environmental Pollution means the accumulation or concentration of wastes that cannot be disposed off by natural recycling process due to their excessive quantity or unique chemical composition [1]. Any substance which is present in nature beyond permissible limits as well as has detrimental effects not only on the environment but also on living organisms is called Pollutant e.g., Cd, Cr, Pb, Zn, Cu, Ni, Ca, and Mg. These chemicals are released into the atmosphere from different natural and anthropogenic sources. High temperature industrial process release coarse fractions of Mg, Ca, Ni, Mn, Cu and Zn. Automobile exhaust and fertilizer industries also release these metals, their compounds, or other salts [2-8]. The urban population is exposed to the aerosol toxic metals that often are well above natural background [9-13]. Many studies on atmospheric metal concentration and their related health hazards have been conducted in several parts of the world which showed diverse fluctuations and disparities among the trace element constituents [14-24]. All these metals produce different diseases like oxides of Zinc along with oxides of Iron produce gastric disorder and vomiting, irritation of skin and mucous

membrane. Nickel, Chromium, Lead, Cadmium, Copper and Carcinogenic calcium causes slowing of heart rate, leukemia and different types of cancer [25-31]. Cobalt and Manganese cause chronic and acute poisoning which results in Anemia and Hypertension [32]. When these chemicals are released into the atmosphere, they enter into the human chain, as soon as they enter biological system cause deaths in some cases. Due to the lack of air quality management capabilities, the Pakistan is suffering from deterioration of air quality. Evidence from various governmental organizations and international agencies has indicated that air pollution is a significant risk to the environment, quality of life and human health. Besides health hazards, heavy metal pollution impair visibility, plays an important role in acidic rain, adversely affects the radiation budget and consequently disturbs a variety of environmental processes may change the cloud properties by nucleation, condensation and chemistry of environment by providing the media for various heterogeneous reactions and carriers for chemical species. Atmospheric aerosol particles are solid or liquid particles suspended in air. Processes that control formation. transformation and the removal of atmospheric aerosols is of great interest in atmospheric science. The reason is that these particles, which are often smaller than 1 micrometer in diameter, play an important part in Earth's radiation budget through the scattering of sunlight and through the interaction with clouds. Human activities, such as burning of fossil fuels and land use, change the properties of the aerosol and

^{*}Address correspondence to this author at the Department of Physics, G.C. University, Faisalabad, Punjab, Pakistan; Tel: 0302-6062879; E-mail: profkhan786@yahoo.com

may therefore influence the climate. This can be either directly through an increase in aerosols or indirectly through the way the anthropogenic aerosols change the way clouds form. Also, heterogeneous reactions on the aerosol particle surfaces influence the gas phase composition and chemistry of the atmosphere. And these particles are responsible for adverse health effects through inhalation. To assess the role of aerosols in our environment and the influence by anthropogenic emissions requires an understanding of the life cycle and transport patterns of solid aerosol particles, their compositional evolution as well as a detailed knowledge of cloud formation and nucleation mechanisms depend on the properties of the preexisting aerosols [33-36].

By definition trace elements are chemical components that naturally occur in soil, plant and wild life in minute concentrations, also known as Trace minerals. They are necessary for the optimal development and metabolic festraining of all living things. For human beings proper cell metabolism effective immune feqrition and healthy reproduction are dependent on a total of 72 of these elements. Since they provide nutritional value, they are sometime effected to as micronutrients. The health giving properties of trace metals vary greatly and some are not even understood yet for instance CV is inversed in the regulation of metabolism of glucose and lipid fats and is also thought to be an aid to weight loss since it promotes the processing of fat for energy rather than its several trace elements are essential storage for the protection, how more regulation and neutrons emission in brain .But when the amount of trace metals becomes greater than permissible limit. They become toxic and generate health hazards.

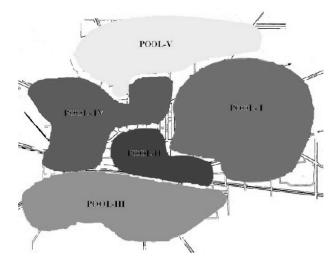


Figure 1: Site selection for Faisalabad City.

Perturbations in signaling is one of the principal modes of action resulting in toxicity which is common to most trace elements. Oxidative processes enhancement resulting in the increase of super oxide anion radical (O^{2-}), H_2O_2 and hydrogen peroxide radical (OH) is the base for other connections with signaling response. Oxidative stress affects numerous cellular components such as DNA, lipids and proteins through oxidation reactions.

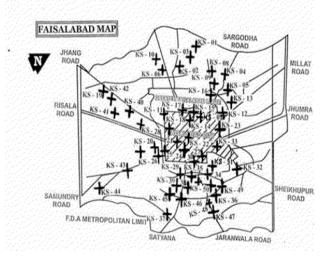


Figure 2: Pool classification for Faisalabad City.

In a very simplistic model of signaling system sleading toward toxicity or tolerance, the following would be the train of events. Trace elements entry in the system is facilitated by transport protein or diffusion. At the topical level there is increased production of H_2O_2 by direct action on NADPH oxidize; at the systemic level there is disruption of phospholipids bilayer due to lipid per oxidation, leading top reduction of ROS inducing synergistic action of SOD, CAT, and APx and increasing H_2O_2 levels especially by SOD. In due time course, excess metals enter cellular organelles like mitochondria and chloroplasts (plants), act as a sink in the electron flow or mist direct the electron flow (depending on the redox status of the metal), which causes production of free radicals. Free radicals, in turn, initiate the antioxidant systems (Halliwel Asada Pathway) to guench H₂O₂. Unquenched H_2O_2 in addition to other free radicals gives rise to singlet oxygen. In addition, receptor metal complex in the plasma membrane causes excess calcium ion concentration which initiates the calmodulin-Ca²⁺ system activating various kinases. These reactive molecules and kinases act as signals on the transcription factors present in the nuclear as well as the organelle DNA, leading to the production of stress protein sand secondary metabolites that can act as

either damage causing agents or stress counting agents.

The present study was conducted in order to assess the concentration of heavy metals in the atmosphere of Faisalabad and their effect on environment. For the confirmation of interactive relationship between solid aerosols and environment co relationship was also established. These results were also compared with other similar studies quoted in national and international journals having impact factors.

2. MATERIALS AND METHODS

In this study attempt has been made to estimate the trace elements like Cd, Cr, Ni, Zn, Cu, Mg and Pb in the atmosphere comprising of various pools of Faisalabad city. 50 sites were randomly selected for analysis covering industrial, transportational, commercial and residential nature of the Faisalabad environment. Air samples containing solid aerosols were collected using Kimoto high volume air sampler from selected areas of Faisalabad. Samples were collected for a period of 12 hrs with an average flow rate of 0.8m³/min. Solid aerosols were trapped on glass fiber filters with the collection efficiency of 90%. The filters were weighed before and after sampling [37]. Then analyzed atomic absorption by spectrophotometer (Model No. Varian AA-1475).

It is compulsory to know the constituents of solid aerosols and their sizes as these parameters not only determine the scattering and absorption properties but also provide us information to interact with gaseous pollutants in the environment and react to form new solid aerosol particles or to modify existing ones by homogeneous, homo molecular nucleation or homogeneous, heterogeneous molecular nucleation and their ability to get into the human respiratory system and generate health hazards.

Atomic absorption spectrometry (AAS) used in this study to work on the principle that the amount of energy absorbed in flame is proportional to the concentration of the element present in the sample. Since each element has its own characteristic absorption wavelength, when the sample is in solution form it is found that the measured extinction coefficient σ is proportional to the concentration of the absorbing substance and hence may be written as

μ= K σ

Where μ is the linear extinction coefficient and K is the extinction coefficient per unit concentration.

The results obtained in this study slightly disagreed from similar previous and current studies conducted by other workers. This disagreement is explained not only in terms of climatological, geological, geographical set ups and latitude, longitude location with respect to solid aerosol sources but also expansion in industrial and transportational set up [36-40].

3. RESULTS AND DISCUSSION

3.1. Elemental Analysis of Solid Aerosols By Atomic Absorption Spectrophotometery (AAS)

In order to determine trace elements through solid aerosols in the Faisalabad. 15 samples were collected from various sites in Faisalabad by dividing it into five pools. After this, all the given samples were subjected to trace elemental analysis by the Atomic Absorption Spectrophotometery (AAS) technique for determination of Ca, Cd, Cr, Ni, Mg, Zn and Pb. The average

Table 1:	Average Concentration of Identified Elements in Selected Pools of Faisalabad Environment
----------	--

Sr. No.	Identification of elements in Solid Aerosols	Pool Selections				
		Pool (1) Average Conc.(ppm)	Pool (2) Average Conc.(ppm)	Pool (3) Average Conc.(ppm)	Pool (4) Average Conc.(ppm)	Pool (5) Average Conc.(ppm)
1	Cadmium (Cd)	2.0942	2.24825	1.95475	2.259	2.596
2	Chromium (Cr)	0.0158	0.0145	0.01125	0.012	0.014
3	Nickel (Ni)	0.4822	0.599	0.35575	0.64275	0.9925
4	Lead (Pb)	1.0324	1.33675	0.2632	0.80575	0.4475
5	Zinc (Zn)	2.1108	2.4635	2.27	2.515	1.38825
6	Magnesium (Mg)	1.1736	1.217	1.206	1.22125	1.15875
7	Calcium (Ca)	8.307	8.30525	3.53875	11.46725	6.7755

Sr. No.	Identification of elements in Solid Aerosols	Pool wise Identification of elements in solid aerosols trend					
		Pool (1) Average Conc.(ppm)	Pool (2) Average Conc.(ppm)	Pool (3) Average Conc.(ppm)	Pool (4) Average Conc.(ppm)	Pool (5) Average Conc.(ppm)	
1	Cadmium (Cd)	P5>P4>P2>P1>P3					
2	Chromium (Cr)	P1>P2>P5>P4>P3					
3	Nickel (Ni)	P5>P4>P2>P1>P3					
4	Lead (Pb)	P2>P1>P4>P5>P3					
5	Zinc (Zn)	P4>P2>P3>P1>P5					
6	Magnesium (Mg)	P4>P2>P3>P1>P5					
7	Calcium (Ca)	P4>P1>P2>P5>P3					

Table 2: Pool Wise Trend Pattern of Identified Elements in Faisalabad Environment

concentration of these trace elements is given in Table 1. The average concentrations of all the elements according to their pools are plotted against sample codes (Cd), (Cr), (Ni), (Cu), (Mg), (Zn) and (Pb). In order to know which region had lighter trace element concentration in solid aerosols loadings than the others. The average pool wise concentration of identified elements were given in Cadmium Cd(1) in amounts ranging between (2.596→1.95475) in Pool $(5\rightarrow 3)$, Chromium Cr(2) in amounts ranging between $(0.0145 \rightarrow 0.01125)$ in Pool $(2 \rightarrow 3)$, Nikel Ni(3) in amounts ranging between (0.9925→0.35575) in Pool $(5\rightarrow 3)$, Lead Pb(4) in amounts ranging between $(1.33675 \rightarrow 0.2632)$ in Pool $(2 \rightarrow 3)$, Zinc Zn(5) in amounts ranging between (2.515→1.38825) in Pool $(4\rightarrow 5)$, Magnesium Mg(6) in amounts ranging between $(1.22125 \rightarrow 1.15875)$ in Pool $(4 \rightarrow 5)$, Calcium Ca(7) in amounts ranging between (11.46725→3.53875) in Pool $(4\rightarrow 3)$ respectively. Following Pool wise Trend pattern of as Identified elements in solid aerosols is given in Table 2.

To check the authenticity of the data and trace metals statistical analysis was carried out which showed the wide variations in SD [41-50] and CV

Zn

Mg

Ca

pointing out the instability of the environment which is not only disturbing the ecology but also the human health (Table 3).

3.2. Health Hazards of Solid Aerosol Samples

Health hazards of heavy metals such as Cd, Cr, Ni, Cu, Mg, Zn and Pb contained in solid aerosols samples selected from different pools were also investigated in this study. These metals were found to be highly toxic and generating serious health hazards like blood pressure, heart attacks, kidney diseases, fibrosis of lungs, constipation and loss of appetite, abdominal pain and paralysis of muscles Gastrointestinal, irritation and vomiting.

The following strategy was adopted in this study.

- Filling of the questionnaires from 500 (250 from Pool1, 100 from Pool 2 and 150 from Pool 3) residents about their health status.
- Interviews of the doctors and effectees of different hospitals and clinics of concerned pools were conducted.

0.426

0.0236

2.996

с٧

28.176 1.355 35.757 50.687

21.824

1.983

39.93

Identified Elements	Мах	Min	Mean	S.D
Cd	2.596	1.95475	2.275	0.641
Cr	0.0158	0.011	0.0134	.0001815
Ni	0.9925	0.35575	0.674	0.241
Pb	1.33675	0.2632	0.801	0.406

1.38825

1.15875

3.53875

1.952

1.19

7.503

Table 3: Statistical Analysis of Major Identified Elements by AAS

2.515

1.22125

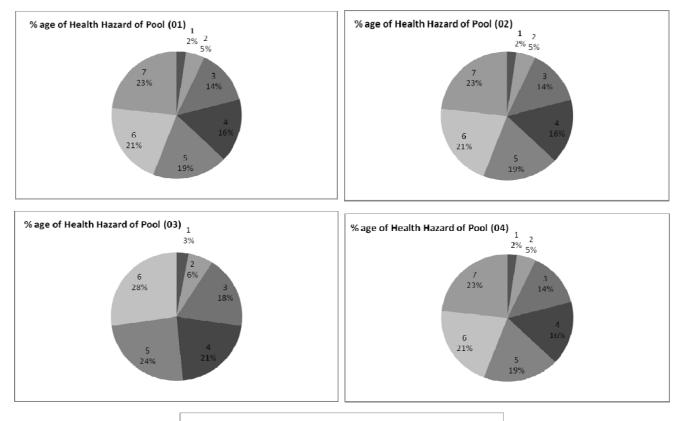
11.46725

- Empirical relations from this data were developed with respect to trace metals.
- Using the empirical relations was found the percentages of the effectees.

The data collected from questionnaires filled by 500 residents indicate that 6.15% had developed cancer, 10.09% ENT diseases, 0.34% respiratory diseases, 2.74% skin gastrointestinal, 33.29% giddiness, 40.47% headache and 6.87% heart attack. Empirical relations were developed employing the statistical method given by Aban Asrar (1996). In these regression equations the values of \mathbb{R}^2 (the coefficient of co-relation) are 87.8 for the heart patients, 68.3 for headache, 66.6 for

giddiness, 79.7 for E.N.T., 73.3 for fatigue, 77.10 for gastrointestinal, 96.5 for respiratory diseases, 69.3 for urinary diseases, 88.7 for blood pressure, 71.5 for anemia, 86.0 for skin diseases and 81.3 for cancer confirmed through personal interviews using these empirical relations the percentage of effectees were found out and their trend was also checked and presented pool wise in Table **4** or from the doctors of some selected clinics existing in the selected areas [51-60].

It is speculated from this study that average health has close linknkage with path ways of solid aerosol population. The elders, infants and persons with chronic, cardiopulmonary diseases, influenza or



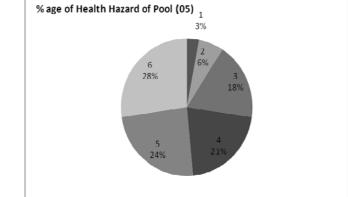


Figure 3: Pool Wise Percentage Health Hazards Related To Faisalabad Environment.

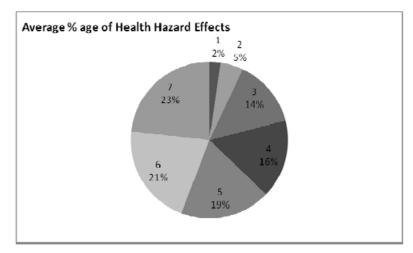


Figure 4: Pi- Graph showing the overall over all average % age of health hazard effects related to total environment of Faisalabad.

Sr. No.	Diseases	Average % Age of Health Hazards Effects	Health Hazards Trend in Faisalabad Environment
1	ENT	10.09002	_
2	Giddiness	33.28468	Head
3	Fatigue	ND	Headaches Attack> C Res
4	Gastrointestinal	ND	
5	Urinary	ND	e> Giddin Cancer> s Spiratory
6	Cancer	6.14492	
7	Heart Attack	6.8685	ess>ENT; Skin Dise Diseases
8	Headache	40.4711	α v
9	Skin Diseases	2.73798	Heart Ses>
10	Respiratory Diseases	0.3381	~

Table 4: Health Hazard's Trend in the Atmosphere of Faisalabad

asthma most susceptible to mortality and serious morbidity effects while others are susceptible to less serious health effects such as transient increase in respiratory symptom, decreased lung function or physio logic changes. These findings were also confirmed through idea of co founding factors and index of agreement, the latest trend used for health implications. Addition knowledge is needed about the specific pollutant from specific area of interest or mixed pollutants from total environment for the adverse health effects and the biologic mechanism involved.

4. CONCLUSIONS

The biotic effects of heavy metals when unduly exposed to them could becomes potentially life threatening hence cannot be neglected, while these metals are in many ways indispensible, good precaution and adequate occupational hygiene should be taken in handling them as per recommendations of the concerned area doctors. Although the heavy metals poising could be clinically diagnosed and medically treated the best option is to prevent heavy metals pollution and the subsequent human poisoning.

ACKNOWLEDGEMENTS

The authors are highly obliged to acknowledge the services of In charge, Central Hi-Tech lab, UAF, along with Dr. Muhammad Ajmal and Dr. Abdul Quddoos with their technical team for providing us Lab facilities, technical assistance when and where needed, their valuable suggestions, in time encouragements, healthy discussions and positive criticism in getting this work completed with utmost ease and perfection. We also express our sincere thanks to all who provided us necessary facilities at different monitoring locations.

REFERENCES

- Othmer K. Encyclopedia of chemical technology. 1st ed. Weily inter Sciences Publication Inc 1978; 624-642.
- [2] Borbely Kiss I, Koltay E, Szabo GY, Bozo L, Tar K. Composition and sources of urban and rural atmospheric aerosol in eastern Hungary. J Aerosol Sci 1999; 30: 369-91. <u>http://dx.doi.org/10.1016/S0021-8502(98)00051-2</u>
- [3] Pakkanen TA, Loukkola K, Korhonen CH, Aurela M, Makela T, Hillamo RE, *et al.* Sources and chemical composition of atmospheric fine and coarse particles in the Helsinki area. Atmospheric Env 2001; 35: 5381-91. http://dx.doi.org/10.1016/S1352-2310(01)00307-7
- [4] Harrison RM, Smith DJT, Pio CA, Castro LM. Comparative receptor modelling study of airborne particulate pollutants in Birmingham (United Kingdom), Coimbra (Portugal) and Lahore (Pakistan). Atmospheric Env 1997; 31: 3309-21. http://dx.doi.org/10.1016/S1352-2310(97)00152-0
- [5] Hien PD, Binh NT, Truong Y, Ngo NT, Sieu LN. Comparative receptor modeling study of TSP, PM2 and PM2-10 in Ho Chi Minh City. Atmospheric Env 2001; 35: 2669-78. <u>http://dx.doi.org/10.1016/S1352-2310(00)00574-4</u>
- [6] Arditsoglou A, Samara C. Levels of total suspended particulate matter and major trace elements in Kosovo, A source identification and apportionment study. Chemosphere 2005; 59: 669-78. <u>http://dx.doi.org/10.1016/j.chemosphere.2004.10.056</u>
- [7] Valavanidis A, Fiotakis K, Vlahogianni T, Bakeas EB, Triantafillaki S, Paraskevopoulou V, *et al.* Characterization of atmospheric particulates, particle bound transition metals and polycyclic aromatic hydrocarbons of urban air in the centre of Athens (Greece). Chemosphere 2006; 65: 760-68. http://dx.doi.org/10.1016/j.chemosphere.2006.03.052
- [8] Jenq FT. Emission of particular matter from three major industries. J Aerosol Sci 1992; 23: 991-94. http://dx.doi.org/10.1016/0021-8502(92)90579-K
- [9] Hadad K, Mehdizadeh S, Sohrabpour M. Impact of different pollutant sources on Shiraz air pollution using SPM elemental analysis. Environment Int 2003; 29: 39-43. http://dx.doi.org/10.1016/S0160-4120(02)00143-5
- [10] Salam A, Bauer H, Kassin K, Ullah SM, Puxbaum H. Aerosol chemical characteristics of a mega city in Southeast Asia (Dhaka-Bangladesh). Atmospheric Env 2003; 37: 2517-28. <u>http://dx.doi.org/10.1016/S1352-2310(03)00135-3</u>
- [11] Samura A, Al-Agha O, Tuncel SG. Study of trace and heavy metals in rural and urban aerosols of Uludag and Bursa (Turkey). Water Air Soil Pollution Foc 2003; 3: 111-29. http://dx.doi.org/10.1023/A:1026053128355
- [12] Zereini F, Alt F, Messerschmidt J, Wiseman C, Feldmann I, Von BA, et al. Concentration and distribution of heavy metals in urban airborne particulate matter in Frankfurt am Main, Germany. Environm Sci Tech 2005; 39: 2983-89. http://dx.doi.org/10.1021/es040040t
- [13] Shridhar V, Khillare PS, Agarwal T, Ray S. Metallic species in ambient particulate matter at rural and urban location of Delhi. J Hazardous Mat 2010; 175: 600-607. <u>http://dx.doi.org/10.1016/i.jhazmat.2009.10.047</u>
- [14] Freitas MC, Pacheco AMG, Verburg TG, Wolterbeek HT. Effect of particulate matter, atmospheric gases, temperature, and humidity on respiratory and circulatory diseases' trends in Lisbon, Portugal. Environm Monit Ass 2010; 162: 113-21.
- [15] Garcia VC, Gego E, Lin S, Pantea C, Rappazzo K, Wootten A, et al. An evaluation of transported pollution and respiratory related hospital admissions in the state of New York. Atmospheric Pollution Res 2011; 2: 9-15. <u>http://dx.doi.org/10.5094/APR.2011.002</u>
- [16] Sohrabpour M, Mirzaee H, Rostami S, Athari M. Elemental concentration of the suspended particulate matter in the air of Tehran. Environm Int 1999; 25: 75-81. <u>http://dx.doi.org/10.1016/S0160-4120(98)00088-9</u>

- [17] Bilos C, Colombo JC, Skorupka CN, Presa MJR. Sources, distribution and variability of airborne trace metals in La Plata city area, Argentina. Environm Pol 2001; 111: 149-58. <u>http://dx.doi.org/10.1016/S0269-7491(99)00328-0</u>
- [18] Rizzio E, Bergamaschi G, Profumo A, Gallorini M. The use of neutron activation analysis for particle size fractionation and chemical characterization of trace elements in urban air particulate matter. J Radioanal Nucl Chem 2001; 248: 21-28. http://dx.doi.org/10.1023/A:1010605519848
- [19] Wang CX, Zhu W, Peng A, Guichreit R. Comparative studies on the concentration of rare earth elements and heavy metals in the atmospheric particulate matter in Beijing, China, and in Delft, the Netherlands. Environm Int 2001; 26: 309-13. http://dx.doi.org/10.1016/S0160-4120(01)00005-8
- [20] Ragosta M, Caggiano R, D'Emilio M, Macchiato M. Source origin and parameters influencing levels of heavy metals in TSP, in an industrial background area of southern Italy. Atmospheric Env 2002; 36: 3071-87. http://dx.doi.org/10.1016/S1352-2310(02)00264-9
- [21] Quiterio SL, da-Silva CRS, Arbilla G, Escaleira V. Metals in airborne particulate matter in the industrial district of Santa Cruz, Rio de Janeiro, in an annual period. Atmospheric Env 2004; 38: 321-31. http://dx.doi.org/10.1016/j.atmosenv.2003.09.017
- [22] Gupta AK, Karar K, Srivastava A. Chemical mass balance source apportionment of PM10 and TSP in residential and industrial sites of an urban region of Kolkata, India. J Hazardous Mats 2007; 142: 279-87. http://dx.doi.org/10.1016/j.jhazmat.2006.08.013
- [23] Hao YC, Guo ZG, Yang ZS, Fang M, Feng JL. Seasonal variations and sources of various elements in the atmospheric aerosols in Qingdao, China. Atmospheric Res 2007; 85: 27-37. http://dx.doi.org/10.1016/j.atmosres.2006.11.001
- [24] Ayrault S, Senhou A, Moskura M, Gaudry A. Atmospheric trace element concentrations in total suspended particles near Paris, France. Atmospheric Env 2010; 44: 3700-707. <u>http://dx.doi.org/10.1016/j.atmosenv.2010.06.035</u>
- [25] Hayes RB. The carcinogenicity of metals in humans. Cancer Causes Con 1997; 8: 371-85. http://dx.doi.org/10.1023/A:1018457305212
- [26] Drasch G, Schopfer J, Schrauzer GN. Selenium/cadmium ratios in human prostates. Biological Trace Element Res 2005; 103: 103-107. http://dx.doi.org/10.1385/BTER:103:2:103
- [27] Stayner L, Smith R, Schnorr T, Lemen R, Thun M. Ann Epi 1993; 3: 114.
 - http://dx.doi.org/10.1016/1047-2797(93)90020-5
- [28] Fanning D. A Mortality Study of Lead Workers, 1926–1985. Archiv Environm Health 1988; 43: 247-51. http://dx.doi.org/10.1080/00039896.1988.9934942
- [29] Selevan SG, Landrigan PJ, Stern FB, Jones JH. Am J Epi 1996; 122: 673.
- [30] Schrauzer GN. Interactive effects of selenium and chromium on mammary tumor development and growth in MMTVinfected female mice and their relevance to human cancer. Biol Trace Element Res 2006; 109: 281-92. <u>http://dx.doi.org/10.1385/BTER:109:3:281</u>
- [31] Singh V, Garg AN. Trace element correlations in the blood of indian women with breast cancer. Biol Trace Element Res 1998; 64: 237-45. <u>http://dx.doi.org/10.1007/BF02783340</u>
- [32] Hammond PB, Beliles RP. Metals In: Toxicology: The Basic Science of Poisons. 2nd ed.
- [33] Andersen ZJ, Wahlin P, Raaschou-Nielsen O, Scheike TS. Ambient particle source apportionment and daily hospital admissions among children and elderly in Copenhagen. Epidemiology 2006; 17: 200-201. <u>http://dx.doi.org/10.1097/00001648-200611001-00510</u>

- [34] Sarnat JA, Marmur A, Klein M, Kim E, Russell AG, Mulholland JA, et al. Associations between source resolved particulate matter and cardiorespiratory emergency department visits. Epidemiology 2006; 17: 267-68. http://dx.doi.org/10.1097/00001648-200611001-00696
- [35] Liu L, Ruddy T, Dalipaj M, Poon R, Szyszkowicz M, You HY, Dales RE, Wheeler AJ. Effects of indoor, outdoor, and personal exposure to particulate air pollution on cardiovascular physiology and systemic mediators in seniors. J Occupat Environm Med 2009; 51: 1088-98. http://dx.doi.org/10.1097/JOM.0b013e3181b35144
- [36] Mavroidis I, Chaloulakou A. Characteristics and expected health implications of annual PM10 concentrations in Athens, Greece. Int J Environm Pol 2010; 41: 124-39. <u>http://dx.doi.org/10.1504/IJEP.2010.032249</u>
- [37] Anil KD. Environmental Chemistry, 3rd ed. Versa-Bharati University, Wiley Eastern Limited Inc 1994; pp. 150-151.
- [38] Perry R, Young RJ. Hand Book of Pollution Analysis, John Weily and Sons, New York Inc 1997; p. 195.
- [39] Kolmer JA, Spaulding EH, Robinson HW. Approved laboratory-techniques, 5th ed. Inc 1959; p. 1089.
- [40] Ahmad S, Daud M, Qureshi IH. Use of biomonitors to assess the atmospheric changes. Proc Pak Acad Sci 2007; 44(3): 201-19.
- [41] Harper HA, Rodwel VW, Mayes PA. Review of physiological chemistry, 16th ed. Inc 1977; pp. 534-540.
- [42] Vercruysse A. Hazardous metals in Human toxicology, 2rd ed. Elsevier Amsterdam. Oxford, New York, Tokyo Inc 1984; pp. 56-62.
- [43] Cholak J. The nature of atmospheric pollution in a number of communities. In National Air Pollution Symposium, Standard Research institute Los Angeles, California 1989; 2nd.
- [44] Facchini H. Heavy metals in air of Milan in the month of Jan. Inst. Fis. Univ. Milan. Italy 9. 865-5, Chem. Abst 1980; 92(23): 18509l w.
- [45] Muthusubramanian P, Deborrah SPM. Estimation of concentration of suspended particulate matter collected in Madurrai city. Ind J Environ Prot 1989; 9(9): 650-54.
- [46] Bowen HJM. Environmental Chemistry of the elements, Academic Press Inc 1979; pp. 6-7.
- [47] Nawaz H. Test your Chemistry, Carwan Printing Press Lahore, Pakistan Inc 2000; 289: 296-97.
- [48] Snedden J. Use of an impaction electro-thermal atomization atomic absorption spectrometric system for direct determination of Cu, Mn, and Cd in the labortary atmosphere. Analytical Lets 1985; 18(A10): 1261-80. <u>http://dx.doi.org/10.1080/00032718508066208</u>

Received on 01-01-2013

Accepted on 28-01-2013

Published on 08-02-2013

http://dx.doi.org/10.6000/1927-5129.2013.09.15

© 2013 Shahid et al.; Licensee Lifescience Global.

This is an open access article licensed under the terms of the Creative Commons Attribution Non-Commercial License (<u>http://creativecommons.org/licenses/by-nc/3.0/</u>) which permits unrestricted, non-commercial use, distribution and reproduction in any medium, provided the work is properly cited.

- [49] Harrison RH, Struges WT. The measurement and interpretation of Br/Pb ratios in airborne particles. Atmos Environ 1983; 17: 311-28. http://dx.doi.org/10.1016/0004-6981(83)90048-3
- [50] Waheed S, Ahmad S, Zaidi JH, Rahman A, Qureshi IH, Saleem M. Transfer of inorganic elements in air and their enrichment in ash during coal combustion. Toxicol Environ Chem 2001; 83: 13-23. <u>http://dx.doi.org/10.1080/716067228</u>
- [51] Daud M, Khalid N, Iqbal J, Ahmad S. Assessment of atmospheric pollution level using Asclepias procera leaves as biomonitor. Radiochim Acta 2007; 95: 423-31. http://dx.doi.org/10.1524/ract.2007.95.7.423
- [52] Doull J, Klaassen CD, Amdur MO, Eds. Macmillan Publishing Co., Inc., New York, NY 2002; vol. 26: pp. 409-467.
- [53] Miskolczi Ferenc M, Mlynczak M. The greenhouse effect and the spectral decomposition of the clear-sky terrestrial adiation. Idojaras Quarterly Journal of the Hungarian 2004; pp. 209-251.
- [54] M Allan R. Combining satellite data and models to estimate cloud radiative effects at the surface and in the atmosphere 2011.
- [55] Spencer RW, Braswell WD. On the Misdiagnosis of Climate Feedbacks from Variations in Earth's Radiant Energy Balance 2011; 3(1): 1603-13.
- [56] McKitrick R, Vogelsang TJ. Multivariate trend comparisons between auto correlated climate series with general trend regressors, Department of Economics, University of McShane 2011.
- [57] Blakely B, Abraham J, Wyner. A Statistical Analysis of Multiple Temperature Proxies: Are Reconstructions of Surface Temperatures Over the Last 1000 Years Reliable? Ann Appl Statist 2011; 5(1): 5-44. http://dx.doi.org/10.1214/10-AOAS398
- [58] GFu Q, Manabe S, Johanson C. On the warming in the tropical upper troposphere: Models vs observations, Geophys Res Lett 2011; 38: L15704, http://dx.doi.org/10.1029/2011GL048101
- [59] OLindzen R, Yong-Sang Choi Y. On the Observational Determination of Climate Sensitivity and Its Implications. Asia-Pacific J Atmos Sci 2011; 47(4): 377-90.
- [60] Ctober-Anderegg William RL, James W, Prall Jacob Haroldand Stephen H. Schneider Expert credibility Change. PNAS 2010; 10: 1073.